## A TOTAL SYNTHESIS OF PARA-FORSSMAN GLYCOLIPID ISOLATED FROM HUMAN ERYTHROCYTE MEMBRANEl)

Shigeki Nunomura, Masato Mori, Yukishige Ito, and Tomoya Ogawa\*

RIKEN (The Institute of Physical and Chemical Research), Wako-shi, Saitama, 351-01 Japan

Abstract: A first total synthesis of Para-Forssman glycolipid, GalNAc $\beta$ 1->3GalNAc $\beta$ 1->3Gal $\alpha$ 1->  $4Ga1\beta1 \rightarrow 4G1c\beta1 \rightarrow 1Cer$ , is described to provide an unambiguous evidence for the proposed structure of the natural product.

A novel globopentaosyl ceramide 1 was isolated from human red blood cells in 1982 and chemically characterized. 1 showed a precipitin reaction with anti-GIoboside I antibody but no reaction with anti-Forssman antibody and was named Para-Forssman glycolipid2. We describe here a first total synthesis of 1, which has provided a synthetic evidence for the proposed structure 1.



Scheme 1 (TMB = 2, 4, 6-trimethylbenzoy), TBDPS =  $Bu'Ph_2Si$ 

Based on a retrosynthetic analysis of 1, a glycopentaosyl glycosyl donor 2 with trimethylbenzoyl auxiliary<sup>3</sup> at O-2a, and two glycosyl acceptors  $3<sup>4</sup>$  and  $4<sup>5</sup>$ as a ceramide equivalents were designed The donor 2 was further disconnected into a glycobiosyl glycosyl donor 5 and a glycotriosyl glycosyl acceptor 6, which were designed as either an imidate 7 or a thioglycoside 8, and a readily available  $9^3$ , respectively.

A practical route to the glycosyl donors 7 and 8 was developed as follows. Silver triflate and powdered molecularsieves 4A (MS4A) promoted glycosylation of alcohol **116** with bromide 106 in  $(CH_2Cl)_2$  gave stereoselectively a 94% vield of  $12<sup>7</sup>$ , which was converted into

147 via 137 in 5 steps (1 4:1 AcOH-H<sub>2</sub>O at 80°, 2 Ac<sub>2</sub>O in Py, 3 MeONa in MeOH, 4 HS(CH<sub>2</sub>)3SH-Et<sub>3</sub>N in MeOH<sup>8</sup>, 5 phthalic anhydride-Et3N-Py then Ac<sub>2</sub>O at 75°, 39% overall). Deallylation of 14 by treatment with  $(\text{Ph}_3\text{Ph}C1\text{-DA}BCO$  in 7:3:1 EtOH-PhH-H<sub>2</sub>O<sup>9</sup> and then HgCl<sub>2</sub>-HgO in 9:1 Me<sub>2</sub>CO-H<sub>2</sub>O gave a 51% yield of  $15<sup>7</sup>$ . Removal of allyl group was achieved more efficiently in the presence of 3:1 PdCl<sub>2</sub>-(Ph<sub>3</sub>P)<sub>3</sub>RhCl in 5:1 AcOH-H<sub>2</sub>O and crude 15 was acetylated to give a 70% yield of 17<sup>7</sup> along with a 27% yield of  $16<sup>7</sup>$  as a by-product. In this specific case addition of (Ph<sub>3</sub>P)<sub>3</sub>RhCl is essential to increase the yield of 17 and in the absence of added (Ph<sub>3</sub>P)<sub>3</sub>RhCl a 1:1 mixture of 16 and 17 was obtained in 80% yield. Treatment of hemiacetal 15 with Cl3CCN-DBU<sup>10</sup> in (ClCH<sub>2</sub>)<sub>2</sub> at -15° afforded a 2:1 mixture of  $\alpha$ - and B-trichloroacctimidates 7 in 68% yield. Another glycosyl



donor 8 was also obtained in 93% yield from 17 by treatment<sup>11</sup> with MeSSnBu<sub>3</sub> and SnCl<sub>4</sub> in  $(CICH<sub>2</sub>)<sub>2</sub>$ .

Crucial couplings between either glycobiosyl donors 7 or 8 and a glycotriosyl acceptor 9 were next examined. TMSOTf-MS4A promoted<sup>12</sup> glycosylation of 9 with 7 in  $(CICH_2)_2$  proceeded quickly at -20' to give a 31% yield of the desired product 19 stereoselectively together with a 65% recovery of 9. Alternatively,  $(Bu<sub>4</sub>N)$ <sub>2</sub>CuBr<sub>4</sub>-AgOTf-MS4A promoted<sup>13</sup> glycosylation of 9 with 8 in CH3NO2 proceeded at 20" to give an 8:l mixture of 19 and 21 in 65% yield, and recovered 9 (31%). Use of  $(CICH_2)_2$  instead of  $CH_3NO_2$  as a solvent deteriorated this reaction to afford a glycal 18<sup>7</sup> (72%) as a major product. The inseparable 8:l mixture of 19 and 21 was separated after convertion into 20 and 22 in 55 and 7% yield, respectively, from 9 in two steps  $(1 \text{ NH}_2\text{NH}_2\cdot\text{H}_2\text{O})$  in EtOH for 14 h at 80°, 2 Ac<sub>2</sub>O-DMAP in Py). Hydrogenolysis of 20 in the presence of 10% Pd-C in 7:3 MeOH-H<sub>2</sub>O followed by acetylation gave an 1:1 mixture of 23<sup>7</sup>  $\alpha$  and  $\beta$  in 94% yield. Treatment of



23 with NH<sub>2</sub>NH<sub>2</sub>·AcOH in DMF14 afforded a 99% yield of 247, which was further converted into glycosyl donors 25 (92%) and 26 (75%, o :  $\beta=1:5$ ) by treatment with CC13CN-DBU in (ClCH<sub>2</sub>)<sub>2</sub> and DAST in  $(CICH<sub>2</sub>)<sub>2</sub><sup>15</sup>$ , respectively.

Crucial glycosylation of 3 with a glycosyl donor 25 was performed in the presence of TMSOTf-MS4A in (ClCH<sub>2</sub>)<sub>2</sub> but the formation of desired product 27 could not be detected by tic examination. Use of a fluoride 26 in the presence of  $SnCl<sub>2</sub>-AgOTf<sup>16</sup>$ -MS4A in (ClCH<sub>2</sub>)<sub>2</sub>, however, did afford a  $6\%$  yield of  $27<sup>7</sup>$ . In order to enhance the efficiency of this crucial coupling, an another glycosyl acceptor 4 was next examined,

## **Scheme 3**

SnCl2-AgOTf-MS4A promoted glycosylation of 4 with 26 augmented the **yield** of the coupled product 28 to 43%. Slightly higher yield of 50% was attained by using  $Cp_2ZrCl_2-AgOTf-MS4A$  as a promotor according to Suzuki et al<sup>17</sup>. 28 was converted via 29 into 27 in two steps (1 Ph<sub>3</sub>P in 1000:1 PhH-H<sub>2</sub>O<sup>18</sup> at 45°, 2 C<sub>23</sub>H<sub>47</sub>COOH and 2-chloro-1-methylpyridinium iodide in 1:1 Bu<sub>3</sub>N-(ClCH<sub>2</sub>)<sup>29</sup>, 98% overall), which was further deprotected to afford the target glycolipid 1 via 30<sup>7</sup> in three steps (1 Bu<sub>4</sub>NF in THF, 2 2:1 0.1M MeONa in MeOH-THF at 20°, 3 2:1 0.25M MeONa in MeOH-THF at  $60^{\circ}$ , overall  $88\%$ ). <sup>1</sup>H-N.m.r. data for synthetic 1 were in full agreement with those<sup>20</sup> for the natural glycolipid.

In conclusion, a stereo-controlled total synthesis of Para-Forssman glycolipid, globopentaosyl ceramide 1, was achieved for the first time by employing glycopentaosyl fluoride 26 as a key glycosyl donor and the proposed structure 1 was eventually confirmed.

*Acknowledgments.* We are grateful to Dr. Susumu Ando for kindly sending us the 500 MHz lHn.m.r. data of natural 1 taken in 49:l (CD3)2SO-D20 at 60°. We thank Dr. J. Uzawa, *Mrs.* T. Chijimatsu, and Mr. K. Fujikura for recording and measuring the NMR spectra and Ms. M. Yoshida and her staff for the elemental analyses. We also thank Ms. A. Takahashi and Ms. K. Moriwaki for their technical assistance.

## Reference and Notes

- *1)* Part *67 in* the series "Synthetic Studies on Cell-Surface Glycans". For part 66, see F. Yamazaki, T. Kitajima. T. Nukada, Y. Ito, and T. Ogawa, submitted for publication.
- 2) S. Ando, K. Ken, Y. Nagai, and T. Yamakawa, New *Vistas in Glycoiipid Res..* Ed. A. Makita, S. Handa, T. Taketomi, and Y. Nagai, Plenum Press, New York, 71 (1982).
- 3) S. Nunomura and T. Ogawa, *Terrahedron Lett., 29 5861 (1988); S.* Sato, S. Nunomura, T. Nakano, Y. Ito, and T. Ogawa, *Tetrahedron Lett., 29 4097 (1988).*
- *4) M. Numata, M.* Sugimoto, S. Shibayama, and T. Ogawa, *Carbohdr. Res., 174 73 (1988).*
- *5)* M. Mori, Y. Ito, and T. Ogawa, *Carbohydr. Res.,* in press; see also P. Zimmermann, R. Bommer, T. Bar, and R. R. Schmidt, J. *Carbohydr. Chem., 7 435 (1988);* K. C. Nicolaou, T. Caulfield, H. Kataoka, and T. Kumazawa, J. Am. Chem. Soc., 110 7910 (1988).
- **6) S.** Sabesan and R. U. Lemieux, Can. J. Chem., 62 644 (1984).
- 7) Physical data for new compounds are compatible with the assigned structures and are given below. Values of  $[\alpha]_D$  and  $\delta_{H,C}$  were measured for the solution in CHCl3 and CDCl3, respectively, at  $25^\circ \pm 3^\circ$  unless noted otherwise. 1:  $\alpha$ ]D -8.0° (c 0.1 py);  $\delta$ H (49:1 (CD3)2SO-D2O, 60") 5.544 (td, 6.7 and 15.3 Hz, Jeer), 5.363 (dd, 6.4 and 15.3 Hz, 4cer), 4.820 (d. 3.6 Hz, lc), 4.560 (d, 8.6 Hz, Id), 4.483 (d, 8.3 Hz. le), 4.267 (d, 7.7 Hz, lb), 4.163 (d. 7.9 Hz, la), I.834 and 1.844 (Zs, 2NAc). and 0.884 (t, 6.7 Hz, 2Me); 7( $\alpha$ ):  $[\alpha]D +49.0^{\circ}$  (c 1.4);  $\delta$ H 8.464 (s, NH), 6.293 (d, 3.6 Hz, 1a); 7(B): ]o]D +30.40" **(C 1.3); 6H** 8.490 (s. NH), 6.260 (d, 8.8 Hz. la); 8: [U]D +24.4" (c 1.2); 6H 5.339 (d, 8.2 Hz, lb), 5.029 (d, 10.4 Hz, la), 2.182, 2.180, 2.082, 2.077, 2.067, and 1.723 (6s AC x 5 and SMe);  $\delta$ C 98.0 (1b), 80.8 (1a); 12:  $\alpha$ ]D -1.7° (c 1.2);  $\delta$ H 5.536 (s, CHPh), 5.512 (d, 8.6 Hz, 1b), 4.255 (d, 7.8 Hz, 1a);  $\delta$ C 101.2 (1a), 100.8 (CHPh), 100.4 (1b); 13:  $\alpha$ ]D +9.0° (c 1.1); 5.454 (d, 2.2 Hz, 4b), 5.447 (d, 8.3 Hz, 1b), 5.319 (d, 2.2 Hz, 4a), 4.213 (d, 7.3 Hz, 1a); 14: [ $\alpha$ ]D +17.8° (c 1.4);  $\delta$ H 5.508 (d, 2.4 Hz, 4b), 5.354 (d, 3.1 Hz, 4a). 5.347 (d, 8.5 Hz, lb), 5.015 (d, 8.5 Hz. la), 4.796 (dd, 3.4 and 11.2 Hz, 3a);

15: [ $\alpha$ ] p +34.8° (c 0.8); 16: [ $\alpha$ ] p -5.0° (c 1.0);  $\delta$ H 5.355 (d, 8.3 Hz, 1b), 4.972 (d, 8.5 Hz, 1a), 4.110 and 4.041 (2d, 17.4 Hz, OCH<sub>2</sub>CO), 2.188, 2.178, 2.076, 2.072, 1.908, and 1.726 (6s, 6Ac); 17: [a]D +27.7° (c 1.6);  $\delta$ H 6.148 (d, 8.8 Hz, 1a), 5.360 (d, 8.2 Hz, 1b), 2.193, 2.178, 2.078, 2.071, 1.875, and 1.722 (6s, 6Ac);  $\delta_C$  98.0 (1b) and 90.4 (1a); 18: [ $\alpha$ ]D +1.7° (c 1.0);  $\delta_H$  6.556 (s, 0.9 Hz, 1a), 5.321 (d, 8.3 Hz, 1b), 2.191, 2.163, 2.101, 2.074, and 1.718 (5s, 5Ac). A mixture of 19 and 21:  $\delta$ H 5.530 (dd, 3.4 and 11.3 Hz, 3e of 19) and 5.660 (dd, 3.4 and 11.3 Hz, 3e of 21); 20: [a]D +12.5° (c 0.6);  $\delta$ H 5.406 (dd, 7.9 and 9.5 Hz, 2a), 5.318 (d, 2.8 Hz, 4e), 5.223 (d, 2.7 Hz, 4d), 5.212 (dd, 3.3 and 11.3 Hz, 3e), 4.954 (d, 3.3 Hz, 1c), 2.270 (s, PhMe), 2.082 (s, 2PhMe), 2.136, 2.033, 2.033, 1.975, 1.889, 1.880 and 1.553 (7s, 7Ac);  $\delta_{\rm C}$  102.9( $^11_{\rm CH}$  163 Hz, 1b), 100.7 ( $^11_{\rm CH}$  160 Hz, 1d), 100.2 ( $^11_{\rm CH}$  163 Hz, 1e), 99.9 ( $^11_{\rm CH}$  160 Hz, 1a) and 99.7 ( ${}^{1}$ J<sub>CH</sub> 168 Hz, 1c); 22: [ $\alpha$ ]<sub>D</sub> +39.3° (c 0.4);  $\delta$ <sub>H</sub> 5.380 (dd, 7.8 and 9.5 Hz, 2a), 2.244 (s, PhMc), 2.089 (s, 2PhMe), 2.079, 2.036, 2.029, 2.001, 1.925, 1.817 and 1.608 (7s, 7Ac);  $\delta_C$  103.4 (<sup>1</sup>J<sub>CH</sub>) 160 Hz, 1b), 99.8 (<sup>1</sup>J<sub>CH</sub> 160 Hz, 1a), 98.5 (<sup>1</sup>J<sub>CH</sub> 160 Hz, 1e), 97.1 (<sup>1</sup>J<sub>CH</sub> 168 Hz, 1c), and 92.6 (<sup>1</sup>J<sub>CH</sub>) 170 Hz, 1d); 23 $\alpha$  and  $\beta$ :  $\delta$ H 6.471 (d, 3.7 Hz, 1a $\alpha$ ) and 5.704 (d, 8.2 Hz, 1a $\beta$ ); 24: [ $\alpha$ ]D +65.8° (c 0.7); 25: [α]<sub>D</sub> +54.6° (c 0.8); δ<sub>H</sub> 8.750 (s, C=NH), 6.836 (s, 2PhH), and 6.698 (d, 3.4 Hz, 1a); 26: α;β=1:5, δ<sub>H</sub> 6.869 (s, 2PhH), 5.893 (dd, 2.4 and 53.4 Hz, 1a $\alpha$ ), 5.489 (dd, 5.2 and 52.5 Hz, 1a $\beta$ ), 5.016 (d, 8.0 Hz, 1c $\beta$ ), 4.989 (d, 8.3 Hz, 1e $\alpha$ ); 27: [ $\alpha$ ]D +36.6° (c 0.2);  $\delta$ H 6.820 (s, 2PhH), 5.012 (d, 8.2 Hz, 1d), 4.946 (d, 3.9 Hz, 1c), 4.790 (d, 8.3 Hz, 1e), 1.009 (s, tBu), and 0.880 (t, 6.7 Hz, 2Me); 28:  $\alpha$ ] $p + 32.3^{\circ}$  (c 0.6);  $\delta H$ 6.831 (s, 2PhH), 5.601 (d, 3.1 Hz, 4c), 5.396 (d, 2.9 Hz, 4d), 5.307 (d, 3.1 Hz, 4e), 4.995 (d, 8.2 Hz, 1d), 4.934 (d, 3.6 Hz, 1c), 4.771 (d, 8.0 Hz, 1e), 4.764 (dd, 1.8 and 11.0 Hz, 2c), 1.051 (s, tBu), and 0.881 (t, 6.7 Hz, Mc); 30: [a]p +14.2° (c 0.1);  $\delta$ H (49:1 (CD3)2SO-D2O, 60°) 6.854 (s, 2PhH), 5.518 (td, 7.1 and 15.4 Hz, 5cer), 5.352 (dd, 6.7 and 15.4 Hz, 4cer), 4.849 (dd, 8.6 and 9.0 Hz, 2a), 4.807 (d, 3.6 Hz, 1c), 4.556 (d, 8.5 Hz, 1d), 4.540 (d, 7.9 Hz, 1a), 4.479 (d, 7.6 Hz, 1c), 4.298 (d, 7.6 Hz, 1b), 2.234 (s, 3PhMe), 1.827 (s, 2Ac), and 0.844 (t, 6.7 Hz, 2Me).

- 8) H. Bayley, D. N. Standring, and J. R. Knowles, Tetrahedron Lett., 19 3633 (1978).
- 9) E. J. Corey and R. Sugg, J. Org. Chem., 38 3224 (1973); P. A. Gent and R. Gigg, J. Chem. Soc. Chem. Commun., 277 (1974), R. Gigg and C. D. Warren, J. Chem. Soc. (C), 1903 (1968).
- 10) R. R. Schmidt and J. Michel, Angew. Chem. Int. Ed. Engl., 19 732 (1980); R. R. Schmidt, ibid., 25 212 (1986).
- 11) T. Ogawa and M. Matsui, Carbohydr. Res., 54 C17 (1977).
- 12) H. Vorbruggen and K. Krolikiewicz, Angew. Chem. Int. Ed. Engl., 14 421 (1975); S. Murata, M. Suzuki, and R. Noyori, Tetrahedron Lett., 2527 (1980); T. Ogawa, K. Beppu, and S. Nakabayashi, Carbohydr. Res., 93 C6 (1981).
- 13) S. Sato, M. Mori, Y. Ito, and T. Ogawa, Carbohydr. Res., 155 C6 (1986).
- 14) G. Excoffier, D. Gagnaire and J.-P. Utille, Carbohydr. Res., 39 368 (1975).
- 15) Wm. Rosenbrook, Jr., D. A. Riley, and P. A. Lartey, Tetrahedron Lett., 26 3 (1985); G. H. Posner and S. R. Haines, ibid., 26 5 (1985).
- 16) T. Mukaiyama, Y. Murai, and S. Shoda, Chem. Lett., 431 (1981); T. Mukaiyama, Y. Hashimoto, and S. Shoda, ibid., 935 (1985).
- 17) T. Matsumoto, H. Maeta, K. Suzuki, and G. Tsuchihashi, Tetrahedron Lett., 29 3567 (1988).
- 18) Yu. G. Gololobov, I. N. Zhmurova, and L. F. Kasukhin, Tetrahedron, 37 437 (1981).
- 19) E. Bald, K. Saigo, and T. Mukaiyama, Chem. Lett., 1163 (1975).
- 20) S. Ando, unpublished results.

(Received in Japan 30 June 1989)